

Bannermanite, a new sodium-potassium vanadate isostructural with $\beta\text{-Na}_x\text{V}_6\text{O}_{15}$ JOHN M. HUGHES¹Department of Earth Sciences, Dartmouth College,
Hanover, New Hampshire 03755

AND LARRY W. FINGER

Geophysical Laboratory, Carnegie Institution of Washington,
Washington, D. C. 20008

Abstract

Bannermanite, $(\text{Na,K})_x\text{V}_x^{4+}\text{V}_{6-x}^{5+}\text{O}_{15}$ ($0.90 > x > 0.54$), a natural vanadium oxide bronze compound, occurs as black subhedral to euhedral crystals in the oxide zone of a fumarole at Izalco volcano, El Salvador, Central America. The mineral is associated with shcherbinaite, stoiberite, and ziesite. Bannermanite is monoclinic, space group $C2/m$, with $a = 15.413(7)$, $b = 3.615(2)$, $c = 10.066(8)\text{\AA}$, $\beta = 109.29(8)^\circ$, $Z = 2$, $D(\text{obs}) = 3.5 \pm 0.2 \text{ g/cm}^3$, and $D(\text{calc}) = 3.55 \text{ g/cm}^3$. The crystal structure of bannermanite has been refined to $wR = 0.074$ from the vanadium oxide bronze structure model. The structure consists of a clinolattice of VO_5 square-based pyramids and VO_6 octahedra. This array of V-O polyhedra forms rectangular tunnels in which Na and K ions occur. The vanadium ions are in two charge states, V^{4+} and V^{5+} , with the number of vanadium ions in the reduced state equal to the number of alkali ions in the tunnel sites; thus, charge balance is maintained.

Bannermanite transmits light on thin edges, with a mean index of refraction of 2.2, on the basis of the rule of Gladstone and Dale. The mineral has two cleavages, $(h0l)$ and $(0kl)$, and a dark gray-black streak. It is named in honor of the late Dr. Harold M. Bannerman.

Introduction

Naturally occurring $(\text{Na,K})_x\text{V}_x^{4+}\text{V}_{6-x}^{5+}\text{O}_{15}$ has been discovered in the fumaroles in the summit crater of Izalco volcano, El Salvador, Central America ($13^\circ 49' \text{N}$; $89^\circ 38' \text{W}$). Identification of this mineral as the potassium-bearing analog of $\text{NaV}_6\text{O}_{15}$ is based on electron probe microanalysis and X-ray diffraction analysis. The new mineral is named bannermanite in honor of the late Dr. Harold M. Bannerman. Dr. Bannerman had a long and distinguished career in the field of geology and served in such posts as Chief of the Non-metalliferous Geology Section of the U.S. Geological Survey, Chief of the Division of Economic Geology of the U.S.G.S., and Professor of Geology, Dartmouth College. The type specimen of bannermanite is currently on deposit at the Department of Earth Sciences, Dartmouth College. Co-type specimens are on deposit at the Smithsonian Institution and the Geological Museum of Harvard University. The total amount of the mineral is a few milligrams. The mineral and mineral name have been approved by the Commission on New Minerals and Mineral Names of the International Mineralogical Association.

Locality and occurrence

Izalco volcano is a basaltic composite cone that has been intermittently active since its birth in 1770. The volcano rises to an altitude of 1965 m, with 650 m relief, and has an approximate volume of 2 km^3 . The geology of the volcano is described by Meyer-Abich (1958), Rose and Stoiber (1969), and Stoiber *et al.* (1975).

Since 1963 the fumarolic gases and sublimate minerals of the summit crater have been extensively studied (Stoiber and Durr 1963; Stoiber and Rose, 1970, 1974; Stoiber *et al.*, 1975; Birnie and Hughes, 1979; Hughes and Birnie, 1980b). Vanadium was found to be a distinctive element in the minerals of the sublimate suite, which includes such vanadium minerals as shcherbinaite ($\text{V}_6^{5+}\text{O}_{15}$), stoiberite ($\text{Cu}_5^{2+}\text{V}_2^{5+}\text{O}_{10}$), and ziesite ($\beta\text{-Cu}_2^{2+}\text{V}_2^{5+}\text{O}_7$).

Bannermanite is found in the MR and L fumaroles (Stoiber *et al.*, 1975) at Izalco volcano as a black incrustation coating basaltic breccia fragments. The mineral occurs as subhedral to euhedral crystals up to approximately $250 \mu\text{m}$ in greatest dimension. It is intimately associated with shcherbinaite.

Introduction to oxide bronzes

The oxide bronzes are compounds of the class M_xTO_n , where M is an electropositive metal ion such as Na or K,

¹ Present address: Department of Geology, Miami University, Oxford, Ohio 45056.

Bannermanite Final Structure Factors

FACTOR = 10.00

H	OBS	CALC	H	OBS	CALC	H	OBS	CALC	H	OBS	CALC	H	OBS	CALC	H	OBS	CALC	H	OBS	CALC	
1	88*	45	-19	520	429	-14	573	566	4	245	244	-4	1486	1488	-20	88*	47	-18	405	430	
3	277	237	-17	343	292	-12	123	131	6	227	225	-2	563	535	-18	625	609	-16	329	337	
H-4	0	0	-15	218	168	-10	1679	1581	6	227	225	0	289	273	-16	368	348	-1	324	214	
2	489	431	-13	675	611	-8	1157	1169	H-3	4	4	2	168	123	-14	368	369	-12	86*	77	
4	277	290	-11	172	156	-6	373	371	4	1076	1035	4	1076	1035	-12	126*	124	-10	410	416	
6	152	91	-9	1046	1020	-4	243	212	4	178	249	6	178	249	-10	492	558	-8	779	776	
8	191	234	-7	1020	1031	-2	119	1031	-15	209	272	8	408	424	-8	571	584	-6	194	196	
10	125*	145	-5	129	152	0	940	851	-13	701	780	10	285	287	-6	38*	48	-4	328	321	
12	176	101	-3	256	231	2	849	843	-11	195	224	12	0*	59	-4	92*	228	-2	269	275	
H-3	0	0	-1	1739	1675	4	227	223	-9	206	160	14	371	319	-2	449	444	0	328	275	
1	104*	35	3	652	627	6	790	743	-7	150	107	H-1	5	5	0	1607	1509	6	187	181	
3	475	466	5	174	41	8	1128	1118	-5	472	434	7	256	315	2	1039	1087	8	126	121	
5	607	618	7	333	361	10	842	819	-3	96*	114	4	225	242	4	225	242	10	424	388	
7	320	334	9	1062	996	12	502*	150	-1	215	224	6	177	184	6	177	184	H-1	8	94	
9	359	446	11	265	262	14	656	621	1	92*	117	-19	180	133	8	441	427	-17	171	130	
11	135*	234	13	529	519	16	201	181	3	381	417	-17	246	240	10	371	331	-15	558	585	
13	440	535	15	282	252	18	76*	60	5	626	749	-15	0*	8	14	351	392	-13	987	1012	
15	117*	99	19	205	188	H-4	3	3	11	214	267	-11	538	547	14	351	392	-11	437	436	
H-2	0	0	-12	56*	121	-10	339	329	H-2	4	4	-7	924	972	-10	39*	4	-11	437	436	
2	786	789	-10	151*	174	-8	377	411	18	443	446	-5	750	680	-8	383	386	-13	987	1012	
4	452	493	-6	377	411	-4	846	822	-16	294	267	-1	212	228	-6	268	252	-7	359*	59	
6	76*	72	-4	215	119	0	215	119	-14	301	228	1	128	17	-4	370	275	-5	186	206	
8	359	431	-2	312	249	2	193	168	-12	147*	90	5	151	115	-2	278	104	-3	356	368	
10	167	259	-10	4	282	273	4	415	482	-10	611	602	7	627	676	9	621	559	1	385	400
12	77*	125	-8	102*	30	6	68*	75	-8	277	272	11	315	315	11	315	315	4	368	401	
14	98*	76	-6	286	277	8	44*	153	-4	627	621	13	394	435	H-3	7	7	5	564	548	
16	51*	30	-4	731	654	10	44*	153	-2	754	751	15	391	357	-15	70*	40	9	516	568	
18	448	525	0	855*	697	H-3	3	3	0	940	893	H-4	6	6	-13	0*	11	11	9	412	429
H-1	0	0	4	2150	2122	4	478	507	2	266	197	H-4	5	5	-11	232	184	H-2	8	8	
1	38*	45	6	2056	1998	-17	478	507	6	348	411	-20	102*	30	-9	59*	91	-20	0*	46	
3	882	895	8	270	271	-15	150	96	8	393	382	-18	413	440	-7	314	295	-18	417	481	
5	844	884	-13	789	808	-13	136	125	10	413	488	-16	165	134	-5	35*	46	-16	403	428	
7	565	621	-11	403	472	-11	403	472	12	261	279	-14	960	874	-3	41*	34	-14	346	248	
9	668	645	-9	216	402	-7	216	145	14	90*	173	-10	59*	59	1	846	893	-12	346	248	
11	378	387	-5	219	184	-5	219	184	H-1	4	4	-8	68*	88	3	194	85	-10	477	485	
13	722	701	-3	77*	60	-3	77*	60	21	192	34	-6	815	836	7	227	240	-8	901	897	
15	178	142	-1	599	582	-1	599	582	6	371	329	-4	2005	2088	9	55	45	-6	256	251	
17	418	461	1	446	460	3	76*	15	-19	571	329	-2	762	755	H-2	7	7	-4	316	350	
19	287	233	5	511	559	-15	511	559	-17	575	565	4	1335	1373	-2	199	178	-2	387	374	
H	0	0	7	150	79	-13	1025	1023	-15	345	341	6	304	292	-18	55*	146	0	365	319	
2	1435	1261	9	150	166	-13	1025	1023	-13	1025	1023	8	566	517	-16	0*	95	4	560	560	
4	794	731	11	163	119	-9	336	322	-11	336	322	10	329	324	-14	509	545	6	201	210	
6	0*	18	9	150	166	-9	336	322	-9	380	223	12	125*	63	-12	765	778	8	109*	128	
8	647	626	-7	21*	97	-7	21*	97	-7	21*	97	14	468	369	-8	555	577	10	370	426	
10	337	348	-5	730	698	-5	730	698	-3	267	276	16	149	134	-8	555	577	12	0*	15	
12	159	134	-3	267	276	-3	267	276	1	184	270	-6	469	467	-4	506	427	H-3	9	9	
14	208	88	H-2	3	3	H-2	3	3	-1	184	270	-4	498*	162	-4	498*	162	-13	239	192	
16	49*	30	-18	295	276	-16	295	276	3	494	626	-3	0*	89	0	238	255	-11	136	194	
			-12	296	249	-12	296	249	5	1035	1130	H-4	6	6	2	112	38	-9	841	808	
			-11	72*	98	-11	72*	98	7	162	17	H-4	6	6	4	490	588				

